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The limiting reactant is the amount of a chemical substance that is consumed, as well as limits how much product forms, during a chemical reaction. In chemistry, calculating the limiting reactant...

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In this case, oxygen is called the “limiting reactant” because it limits how much propane can react. Notice that the limiting reactant isn’t always the substance that is present in the fewest number of moles. In this example, propane (C₃H₈) is the “excess reactant” because after the reaction there will be some of it left over.

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