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The hybridization of the central carbon in $\text{CH}_3\text{C}\equiv\text{N}$ and

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the bond angle CCN are a. sp^2 , 180° . b. sp , 180° . c. sp^2 , 120° . d. sp^3 , 109° . 2. Which of the following statements about an sp hybridized carbon is FALSE? a. It is divalent. b. It forms bonds that are linear. c. It has two p orbitals. d.

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