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calculate  $K$  based on the earth measurement  
then  $g_{\text{Mars}}$  based on spring measurement  
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K Wact = 125 mol The solution is Tf =  
549.39K c J mol K solution The actual is  
work 499.14K. is 25% Then greater h (b)  
Repeat the calculation with a temperature-  
dependent heat capacity  $10 - 2 T -$   
 $3499 \quad 10 - 5 T^2 + 7.464 \quad 10 - 9 T^3 \dots$   
CP (T) = 22.243 + 5977 Assuming

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reversibility  $T_f = 479.44\text{K}$ .  
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