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GCC CHM130 Diebolt Chapter 11 Homework Answers 2011 1 of 3 CHM 130: Chapter 11 Homework Answer Key - errors corrected 1. Check all of the following that are properties of gases: b) indefinite shape d) can diffuse uniformly within same container e) can expand f) can compress g) have very low densities, about 1000 times less dense than water 2.

~~Chapter 11 Homework Answer Key - CHM 130 Chapter 11 ...~~

CHAPTER 11 REVIEW Gases Class SHORT ANSWER Answer the following questions in the space provided. c c The molar mass of a gas at STP is the density of that gas (a) multiplied by the mass of 1 mol. (b) divided by the mass of 1 mol. nRT (c) multiplied by 22.4 L. (d) divided by 22.4 L. For the expression $V =$ (a) increasing P (b) decreasing T

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molesCO₂ 6 mol CO₂ = moles glucose 1 molglucose. The number of moles of CO₂ produced is thus. molesCO₂ = molglucose × 6 molCO₂ 1mol glucose = 0.251 mol glucose × 6 mol CO₂ 1 molglucose = 1.51 mol CO₂. 3. Use the molar mass of CO₂ (44.010 g/mol) to calculate the mass of CO₂ corresponding to 1.51 mol of CO₂:

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Answer: Redox reaction; reductant is Al, oxidant is H₂O; Al is oxidized, H is reduced. This is the reaction that occurs when Drano is used to clear a clogged drain. Redox reaction; reductant is Mg, oxidant is TiCl₄; Mg is oxidized, Ti is reduced. Exchange reaction.

~~Chapter 11.6: Types of Chemical Reactions - Chemistry ...~~

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