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RNA (Updated)

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vs. Meiosis: Side by

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Pogil 1. Name 3
strong acids and write
their formulas. What
makes an acid
strong? +hydrochloric

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acid HCl, nitric acid

HNO_3 , sulfuric acid

H_2SO_4 . Strong

means 100% ionized,

all the H ions have

broken off the acid

molecule in water. It

exists as all ions. 2.

Name 4 weak acids

and write their

formulas.

Strong Versus Weak

Acids Worksheet

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Biology 5. All the internal structures are suspended (floating) in what substance? 6. One of the bacteria in Model 1 has a tail-like structure. a. What is this structure called? b. What might be the purpose of this structure? c. Based on your answer to the previous question,

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PDF Answer

what might you infer
about the ...

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Weak Acids

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Hydrochloric acid is a strong acid - virtually 100% ionised Each mole of HCl reacts with the water to give 1 mole of hydrogen ions and 1 mole of chloride ions That means that if the concentration of

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HNO₃ – Nitric Acid

H₂SO₄ – Sulfuric

Acid HClO₄ –

Perchloric Acid Weak
acids do not

dissociate completely.

Two common (AP
questions) weak acids
are: HF –

Hydrofluoric Acid HF

(aq) + H₂O (l) ⇌ H₃

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Acids, Strong Bases,
Weak Bases Acids ...
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Strong Vs Weak Acids

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with strong acids,

releasing carbon

dioxide: $\text{CaCO}_3(\text{s}) + 2$

$\text{HCl}(\text{aq}) \rightarrow \text{CaCl}_2$

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